Determination of the Concentrations of Heavy Metals in Borehole Water near Refuse Dump Sites around the University of Cross River State (UNICROSS) Calabar, Nigeria

Nwuyi Okori Sam-Uket a, Elvis Monfung Ayim a*, Samuel Monday Ameh b and Muhammed Harun Isah b

a Department of Animal and Environmental Biology, Faculty of Biological Sciences, University of Cross River State, Calabar, Nigeria.

b Department of Biology, Faculty of Science, Confluence University of Science and Technology, Osara, Kogi State, Nigeria.

Authors’ contributions

This work was carried out in collaboration among all authors. All authors read and approved the final manuscript.

ABSTRACT

The contamination of groundwater from the infiltration of leachates into aquifers and health risks on residents drinking from boreholes around dumpsites informed this study on the levels of heavy metals in the boreholes around the University of Cross River State (UNICROSS) dumpsite. Borehole water samples were aseptically collected from 3 boreholes at 50, 80, and 200 meters respectively from the dumpsite into a one-liter sampling bottle and preserved in an ice chest. The
samples were then analyzed for heavy metals using Atomic Absorption Spectrophotometer. The values obtained were compared with WHO permissible standards for drinking water. The mean concentrations of Pb (0.614 ± 0.008), Cd (0.399 ± 0.053), As (0.023 ± 0.002), and Co (0.002 ± 0.001 mg/l) in the borehole water were significantly higher than WHO permissible limits, while the concentrations of Cr (0.002 ± 0.001) and Ni (0.056 ± 0.004) were significantly lower than the WHO permissible limits. The increasing trend of heavy metals in the borehole water was Cr = Co < As < Ni < Cd < Pb. Due to the health risks associated with drinking water contaminated with heavy metals, we recommend that Government should relocate the dumpsite far away from its current location.

Keywords: Dump sites; borehole; heavy metals; solid waste.

1. INTRODUCTION

Water is pervasive and drinking water of good quality is crucial to human physiology and the presence of man relies significantly upon the accessibility of water. A typical man (53 kg - 63 kg body weight), needs around 3 liters of water and food day to day to keep up with great well-being [1] and consequently, water is depicted as the most irreplaceable substance throughout everyday life [2]. In certain spots, accessibility of water is basic, restricted, and sustainable. A deficiency of water could prompt sickness flare-ups and financial misfortune, subsequently, water is a need; it is a special fluid and without it life is unimaginable. Water assumes an essential part in the legitimate working of the world's environment. Man involves water for different purposes which incorporate drinking, transportation, modern and homegrown use, water system of farming, entertainment, fisheries, and garbage removal, quarrying, among others [3,4]. In emerging nations, the human populace increment has prompted a colossal strain on the arrangement of safe drinking water [5]. Perilous water is a worldwide general well-being danger, causing diarrheal and different sicknesses [6]. The majority of the water bodies all around the world are getting polluted accordingly diminishing the use of water for different purposes [7].

An open waste dump is a land disposal site through which all types of waste are disposed-off in a manner that does not protect the environment. They breed vectors of disease, reduce the aesthetic values of the environment, cause nuisance, and produce leachate which infiltrates into the hydrogeological system [8]. Different Sources such as garbage, spoiit foodstuffs, electronic goods, painting waste, and used batteries among others, when dumped with municipal solid wastes raise the concentration of heavy metals in dumpsites. Dumping devoid of the separation of hazardous waste can further elevate noxious environmental effects [8]. The environmental impact of land filling of municipal solid waste can usually result from the run-off of the toxic compounds into surface water and groundwater [8], which eventually leads to water pollution as a result of percolation of leachate [9]. These metropolitan waste unloading areas are assigned spots not as expectedly planned or developed. Subsequently, squanders unloaded throughout the years bio-degenerate and produce leachates that eventually become point wellspring of contamination into soil and groundwater [9]. At the point when precipitation happens, the permeating water (leachates) disintegrates numerous natural and inorganic salts which might be shipped to local springs bringing about the adjustment of the water quality. The consequences of the dumpsite on groundwater hydrology are those leachates from the dumpsite penetrates the ground and move towards groundwater stream subsequently defiling the groundwater along its way [10,11].

Underground water contamination has turned into a worry in Nigeria, because of expanded urbanization and industrialization. Squander released in dump destinations contains a ton of metals, synthetic substances, and miniature life forms which are fit for contaminating ground waters. The UNICROSS dump site is found near residential areas of the institution and it goes under pressure from the consistent dumping of waste from the inhabitants. In addition, the major erosion control channel passes through the institution which most times causes floods within the university community during heavy downpours. The fears of the underground leaking of waste to the environment and its potential impact on human health at large is the reason for undertaking this study to assess the possible impact of solid waste dumps on groundwater quality and ascertain the current water quality status of boreholes water in the vicinity of UNICROSS community to specifically
evaluate the concentration of heavy metal (Hg, Pb, Cr, Cd, As, Ni, Co) and to examine how safe drinking of the borehole water around the dump site is, by comparing its heavy metal levels with the WHO acceptable limits.

2. MATERIALS AND METHODS

2.1 Study Area

The study area is situated at the University of Cross River State, Calabar, between longitude 8°15' E and 8° 25' E and latitude 4°84'N and 4° 54' N under Calabar-South Local Government Area, near the popular Ekpo-Abasi junction. The area is located in the sub-equatorial belt characterized by the wet and dry seasons. The wet season starts in April and finishes in November with a peak in June and July, while the dry season begins in December and finishes in March [12]. The vegetation is notwithstanding, impacted by exercises like farming, development, and urbanization. Aside from the wind system, other climatic boundaries, for example, mean yearly precipitation and temperature, worldwide radiation, and reflection coefficient etc., likewise, impact the region [13]. Three boreholes were chosen for the evaluation. Borehole 1 is found 50 meters from the dumpsite, borehole 2 is 80 meters from the dumpsite, and borehole 3 is 200 meters from the dumpsite. The WHO acceptable limits of every heavy metal were assigned as the control.

2.2 Collection of Water Samples for Heavy Metal Analysis

The technique for sample collection and analysis was espoused by America Public Health Association [14]. Water samples (n = 9) for the study were collected from three boreholes around the study area for three (3) months (January, June, and November 2021), covering both wet and dry seasons.

The samples were aseptically collected. The mouth of the tap was cleaned with methylated spirit using cotton wool and flushed before the collection of samples. The water samples were collected into sterilized sampling bottles, sterilized with 99% ethanol soaked in cotton wool. The water was allowed to run to waste for about 3 to 5 minutes before collection and the water samples were collected into sterile 500 ml bottles and covered tightly. The water samples were then preserved in an ice chest immediately after collection, before transporting to the Ministry of Science and Technology Laboratory, Uyo for analysis of mercury, lead, chromium, cadmium, arsenic, nickel, and cobalt utilizing an Atomic Absorption Spectrophotometer.

2.3 Analysis of Water for Heavy Metals

Heavy metals concentration in water was determined using an atomic absorption spectrophotometer (Perkin Elmer, 2280 model). To 50 ml of unfiltered water samples in a 500 ml Taylor flask, 0.51 ml of concentrated sulphuric acid was added. This was boiled to get white fumes, and allowed to cool, and 1.0 ml of 60% HClO₃ and 5.0ml of concentrated HNO₃ were then added. The resulting mixture was then digested until a clear digest is obtained. This digest was allowed to cool, before being filtered using No 44 Watt man paper into 500 ml volumetric flask, and the digest was analyzed using Atomic Absorption Spectrophotometer for heavy metals in mg/L [15].

2.4 Statistical Analysis

Data collected were subjected to descriptive statistics (mean, standard deviation, and ranges). Analysis of variance (ANOVA) was used to test for the level of the significant difference in the heavy metals concentration between each borehole water compared to the control at 0.05 level of significance and their relevant degrees of freedom. All statistical analysis was carried out using Predictive analytical software (PASW) version 20.

3. RESULTS AND DISCUSSIONS

The results of the concentration of heavy metals in borehole water near the UNICROSS dumpsite are presented in Table 1 below. The highest concentrations of all the heavy metals were recorded in borehole 1, while the least concentrations were recorded in borehole 3. The mean concentration of lead, cadmium, arsenic, and cobalt (borehole 1 and borehole 2) in each borehole was significantly higher than the control (p<0.05), while the concentrations of chromium and nickel were significantly lower than the control (p<0.05). Mercury was not detected in all boreholes during the study. The mean concentrations of heavy metals in the water from boreholes under study are 0.614 ± 0.008, 0.002 ± 0.001, 0.399 ± 0.053, 0.023 ± 0.002, 0.056 ± 0.004, 0.002 ± 0.001 mg/l for lead, chromium, cadmium, arsenic, nickel, and cobalt respectively. The heavy metals concentration in water from the boreholes had an increasing trend of Cr = Co < As < Ni < Cd < Pb.
Table 1. The concentration of heavy metals in borehole water around UNICROSS dumpsite during the study

<table>
<thead>
<tr>
<th>Heavy metals (mg/l)</th>
<th>Control (WHO, 2006)</th>
<th>Boreholes around UNICROSS dumpsite</th>
<th>Mean</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td></td>
<td>Borehole 1</td>
<td>Borehole 2</td>
</tr>
<tr>
<td>Hg</td>
<td>-</td>
<td>BDL</td>
<td>BDL</td>
</tr>
<tr>
<td>Pb</td>
<td>0.01&lt;sup&gt;a&lt;/sup&gt;</td>
<td>0.622 ± 0.001&lt;sup&gt;b&lt;/sup&gt;</td>
<td>0.611 ± 0.008&lt;sup&gt;b&lt;/sup&gt;</td>
</tr>
<tr>
<td></td>
<td>(0.622 – 0.623)</td>
<td>(0.604 – 0.620)</td>
<td>(0.600 – 0.615)</td>
</tr>
<tr>
<td>Cr</td>
<td>0.05&lt;sup&gt;b&lt;/sup&gt;</td>
<td>0.002 ± 0.000&lt;sup&gt;b&lt;/sup&gt;</td>
<td>0.001 ± 0.001&lt;sup&gt;b&lt;/sup&gt;</td>
</tr>
<tr>
<td></td>
<td>(0.002 – 0.002)</td>
<td>(0.001 – 0.002)</td>
<td></td>
</tr>
<tr>
<td>Cd</td>
<td>0.003&lt;sup&gt;a&lt;/sup&gt;</td>
<td>0.433 ± 0.002&lt;sup&gt;b&lt;/sup&gt;</td>
<td>0.389 ± 0.070&lt;sup&gt;b&lt;/sup&gt;</td>
</tr>
<tr>
<td></td>
<td>(0.432 – 0.435)</td>
<td>(0.308 – 0.433)</td>
<td>(0.304 – 0.416)</td>
</tr>
<tr>
<td>As</td>
<td>0.01&lt;sup&gt;a&lt;/sup&gt;</td>
<td>0.026 ± 0.001&lt;sup&gt;b&lt;/sup&gt;</td>
<td>0.024 ± 0.001&lt;sup&gt;b&lt;/sup&gt;</td>
</tr>
<tr>
<td></td>
<td>(0.025 – 0.027)</td>
<td>(0.023 – 0.025)</td>
<td>(0.020 – 0.022)</td>
</tr>
<tr>
<td>Ni</td>
<td>0.07&lt;sup&gt;a&lt;/sup&gt;</td>
<td>0.061 ± 0.001&lt;sup&gt;b&lt;/sup&gt;</td>
<td>0.056 ± 0.004&lt;sup&gt;b&lt;/sup&gt;</td>
</tr>
<tr>
<td></td>
<td>(0.060 – 0.062)</td>
<td>(0.052 – 0.060)</td>
<td>(0.050 – 0.054)</td>
</tr>
<tr>
<td>Co</td>
<td>0.001&lt;sup&gt;a&lt;/sup&gt;</td>
<td>0.003 ± 0.000&lt;sup&gt;b&lt;/sup&gt;</td>
<td>0.002 ± 0.001&lt;sup&gt;b&lt;/sup&gt;</td>
</tr>
<tr>
<td></td>
<td>(0.002 – 0.003)</td>
<td>(0.002 – 0.003)</td>
<td>(0.001 – 0.002)</td>
</tr>
</tbody>
</table>

Values are in mean ± standard deviation, and ranges are in parenthesis ( ).

Values with different superscripts in each borehole compared to the control are significantly different (p<0.05).
3.1 Discussion

The disposal of wastes in the UNICROSS dumpsite poses a potential health danger to people and can pollute the borehole water situated around the dumpsite because of uncontrolled penetration of leachate from the squanders [16]. Essentially, the deterioration of most waste produces methane, which is equipped for causing fire flare-ups, green house emanation as well as the development of solid leachate, which dirty surface and groundwater assets [17]. To this end, this study examines, interestingly the degrees of heavy metals in borehole water from boreholes near dumpsites in UNICROSS, to anticipate the potential health risks of drinking from these boreholes, essentially because these boreholes are the only source of water that residents of this locality rely on for their daily activities such as; drinking, washing items, irrigation water for all the vegetable farms scattered around the institution. Assessing the quality of this water by comparing heavy metal concentrations with international water quality standards is, therefore, necessary to safeguard human life.

This study uncovered minor variations in the levels of heavy metals between the boreholes near the dump site. This could be because of the distinction in the amount of leachates that penetrated the ground water [18]. The slight variation in the concentration of heavy metals between the boreholes as observed was not statistically significant at (P>0.05), however, cobalt (Co) in borehole 3 differed statistically from borehole 1 and 2 (p<0.05) and could be because of the distinction in the depth of boreholes, the distance of the boreholes from the dumpsite, the unearthing strategy utilized in penetrating the boreholes, and the hydrological soil layers [19,11], and structure of waste arranged [20], nonetheless, the concentrations of heavy metals did not vary considerably with the distance from the dumpsites. The concentrations of lead, cadmium, arsenic, and cobalt (boreholes 1 and 2) in the boreholes around the dumpsites were higher than that of the control. This denotes that the levels of the aforementioned heavy metals in boreholes around the dumpsite were raised by the infiltration of contaminants and leachates of the dumpsites. The higher concentration of aforementioned heavy metals in the boreholes could be attributed to the solid waste disposed in the dumpsite which over time biodegrade and add their metallic content to the soil [21]. It could also be a result of the ability of dumpsites to transfer significant levels of toxic and persistent metals into the soil environment, which thereafter infiltrate into groundwater [22]. From this study, borehole 1 had the highest concentration of lead, chromium, cadmium, arsenic, nickel, and cobalt, and this could be a result of its closeness to the dumpsite and as such receive a higher quantity of leachates [23].

The mean cadmium, lead, chromium, nickel, and cobalt concentration in the boreholes examined were lower than the findings from similar studies on the pollution status of heavy metals in groundwater systems around open dumpsites in Abakaliki urban [8] the heavy metals concentration in Uyo metropolis [24], the threats of contaminants in boreholes along Ikot Effanga dumpsite, Calabar Municipality, Nigeria [25]. The variations in the levels of heavy metals between the boreholes for the present study and those of the other studies compared could be due to the difference in age of the dumpsites, depth of the groundwater, its distance from the dumpsite, excavations technique used in drilling the boreholes and the hydrological soil strata. It could also be a result of differences in composition of waste and geographical location. Underground water contamination is the function of waste management strategies and seasons [20]. The differences in the contamination level of underground water could also be due to the differences in leachate percolation, chemical composition of leachate, rainfall, depth and distance of the boreholes from the dumpsite [26,27]. Lead was the highest contaminant in the borehole waters around the UNICROSS dumpsite, and this denotes that the composition of the waste commonly deposited in the dumpsite is dominated by lead. The mean concentration of lead, cadmium, arsenic, and cobalt in the boreholes around the UNICROSS dumpsite was above the WHO acceptable limit for drinking water indicating unsafe levels capable of causing severe health issues to consumers. For example, the ingestion of Arsenic at unsafe levels could cause skin, lung, bladder, and kidney disorders. Lead ingested at an unhealthy level could cause neurological effects, hypertension, impaired fertility, and anaemia. Unsafe levels of cadmium and cobalt could cause kidney disorders [28].

4. CONCLUSION

Solid waste disposed in exposed dumpsite is usually subjected to a series of complex biochemical and physical processes, which lead
to the production of both leachate and gaseous emissions. When leachate leaves the dumpsite and reaches the water table, it results in borehole contamination. This study revealed that the citing of boreholes around the UNICROSS dumpsite could pose a serious health threat to residents. To this end, it is recommended that Government moves the dumpsite far away from its current location to prevent the contamination of groundwater and a possible outbreak of various water-borne diseases and death. Also, Government should insist on consistent treatment of borehole waters by owners to prevent any possible outbreak of diseases.

COMPETING INTERESTS
Authors have declared that no competing interests exist.

REFERENCES
13. Monechot WO. Leachate, groundwater, surface stream, treated water and soil characteristics of the vicinity of a municipal solid waste dumpsite at Uyo metropolis, Akwa – Ibom state, Nigeria. A project submitted in partial fulfillment of the requirement for the award of the Master of science degree (MSC) in analytical chemistry in the Department of pure and industrial chemistry, Faculty of Physical Sciences, University of Nigeria Nsukka; 2009.


© 2023 Sam-Uket et al.; This is an Open Access article distributed under the terms of the Creative Commons Attribution License (http://creativecommons.org/licenses/by/4.0), which permits unrestricted use, distribution, and reproduction in any medium, provided the original work is properly cited.

Peer-review history:
The peer review history for this paper can be accessed here:
https://www.sdiarticle5.com/review-history/101802